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REVIEW: Naming Ionic and Covalent Compounds

Short Answer

1. Name PF_3
2. Name N_2O_3
3. Name: OF_2
4. Name NCl_3
5. Name: SiO_2
6. Name: PCl_5
7. Name: $\text{Mg}(\text{NO}_2)_2$
8. Name: FeBr_3
9. Name: $\text{Cr}(\text{NO}_3)_3$
10. Name: $\text{Cu}_3(\text{PO}_4)_2$
11. Write the formula for tetranitrogen monoxide
12. Write the formula for silicon tetrachloride
13. Write the formula for dinitrogen tetroxide.
14. Write the formula for tetraphosphorous hexoxide.
15. Write the formula for phosphorous penasulfide
16. Write the formula for bromine pentafluoride
17. Write the formula for chlorine trifluoride
18. Write the formula for silicon monosulfide
19. Write the formula for potassium nitrate, show the cross-down.
20. Write the formula for copper(II)sulfate show the cross-down.